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## Structure Reports

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## N-Phenyl-2-(phenyliminomethyl)pyrrole-1-carboxamide

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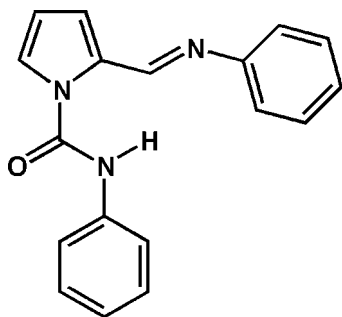
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Key indicators: single-crystal X-ray study;  $T = 183$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.041;  $wR$  factor = 0.082; data-to-parameter ratio = 16.4.

The title compound,  $\text{C}_{18}\text{H}_{15}\text{N}_3\text{O}$ , was prepared from phenyl-(1*H*-pyrrol-2-ylmethylene)amine and phenyl isocyanate in the presence of catalytic amounts of  $[\text{Pd}(\text{PPh}_3)_4]$ . The conformation of the molecular structure is determined by an intramolecular hydrogen bond between the amide NH function and the imine N atom. The molecule is essentially planar. Only the peripheral phenyl substituents are bent out of the plane.

## Related literature

For related literature, see: Mishriky *et al.* (1998).

## Experimental

## Crystal data

$\text{C}_{18}\text{H}_{15}\text{N}_3\text{O}$	$V = 2866.1$ (4) Å <sup>3</sup>
$M_r = 289.33$	$Z = 8$
Orthorhombic, <i>Pccn</i>	Mo $K\alpha$ radiation
$a = 18.292$ (2) Å	$\mu = 0.09$ mm <sup>-1</sup>
$b = 19.966$ (2) Å	$T = 183$ (2) K
$c = 7.8479$ (4) Å	$0.15 \times 0.1 \times 0.1$ mm

## Data collection

Nonius KappaCCD diffractometer	3270 independent reflections
Absorption correction: none	1391 reflections with $I > 2\sigma(I)$
11352 measured reflections	$R_{\text{int}} = 0.082$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$	199 parameters
$wR(F^2) = 0.082$	H-atom parameters constrained
$S = 0.74$	$\Delta\rho_{\text{max}} = 0.14$ e Å <sup>-3</sup>
3270 reflections	$\Delta\rho_{\text{min}} = -0.20$ e Å <sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N3}-\text{H3N}\cdots\text{N2}$	0.88	1.83	2.691 (2)	166

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *DENZO* (Otwinowski & Minor, 1997); data reduction: *DENZO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1990); software used to prepare material for publication: *SHELXL97* and *XP*.

The author gratefully acknowledges financial support by the Deutsche Forschungsgemeinschaft (SFB 436).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2518).

## References

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**supplementary materials**

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## ***N*-Phenyl-2-(phenyliminomethyl)pyrrole-1-carboxamide**

**W. Imhof**

### **Comment**

Derivatives of the title compound have only been described once as intermediates in the synthesis of 1*H*-pyrrolo[1,2-*c*]imidazoles which were synthesized because of their molluscicidal activity (Mishriky *et al.*, 1998). To the best of our knowledge none of these derivatives have been structurally characterized.

The title compound is produced by the formal insertion of an isocyanate into the N—H bond of the pyrrol ring. The conformation of the amide and the imine substituent relative to each other is determined by an intramolecular N3—H3N $\cdots$ N2 hydrogen bond between the amide NH function and the imine nitrogen atom. The other bond lengths and angles are of expected values.

### **Experimental**

230 mg (1.35 mmol) Phenyl-(1*H*-pyrrol-2-ylmethylene)-amine and 240 mg (2.025 mmol) phenylisocyanate were refluxed in 20 ml of THF together with 65 mg (0.056 mmol) [Pd(PPh<sub>3</sub>)<sub>4</sub>] and 4 mg glacial acetic acid for 2 hrs. Evaporation of the solvent yielded an orange oil. Column chromatography on silica yielded 90 mg (23%) of the title compound using a mixture of pentane and ethyl acetate (5:1) as the eluent. With a ratio of 2:1 a compound which corresponding to it's mass spectrum is composed of one equivalent of imine and two equivalents of isocyanate was obtained (40 mg, 7%). Colorless crystals of the title compound were produced from a solution in a pentane/ethyl acetate mixture (10:1) at -20 °.

MS (EI) [*m/z* (%): 289 (*M*<sup>+</sup>, 4), 197 (C<sub>12</sub>H<sub>9</sub>N<sub>2</sub>O<sup>+</sup>, 40), 169 (C<sub>11</sub>H<sub>9</sub>N<sub>2</sub><sup>+</sup>, 100), 119 (PhNCO<sup>+</sup>, 76), 91 (C<sub>7</sub>H<sub>7</sub><sup>+</sup>, 41), 77 (C<sub>6</sub>H<sub>5</sub><sup>+</sup>, 28), 64 (C<sub>5</sub>H<sub>4</sub><sup>+</sup>, 16), 51 (C<sub>4</sub>H<sub>3</sub><sup>+</sup>, 15), 39 (C<sub>3</sub>H<sub>3</sub><sup>+</sup>, 10); <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 293 K) [p.p.m.]: 6.44–6.48 (m, 2 H, CH<sub>pyrrole</sub>), 7.01–7.05 (m, 1H, CH<sub>pyrrole</sub>), 7.18–7.77 (m, 8H, CH<sub>ar</sub>), 8.21–8.23 (m, 2H, CH<sub>ar</sub>), 8.38 (s, 1H, CH<sub>imine</sub>), 14.74 (s, 1H, CH<sub>amide</sub>); <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 293 K) [p.p.m.]: 110.7, 120.7, 120.9, 124.1, 126.6, 128.1, 129.0, 129.6, 130.3, 131.4, 138.4, 148.2, 148.8, 151.3.

### **Refinement**

Hydrogen atoms were calculated in idealized positions and refined with distances of 0.88 Å (N3—H3N) and 0.95 Å (C—H). All hydrogen atoms were refined using a riding model with *U*<sub>iso</sub>(H) = 1.5 times *U*<sub>iso</sub>(C, N).

## Figures

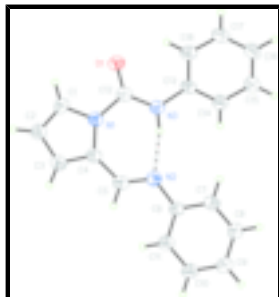


Fig. 1. The molecular structure of the title compound showing the labelling scheme. Displacement ellipsoids are presented at the 40% probability level. The dashed line indicates a hydrogen bond.

## *N*-Phenyl-2-(phenyliminomethyl)pyrrole-1-carboxamide

### Crystal data

$C_{18}H_{15}N_3O$

$M_r = 289.33$

Orthorhombic, *Pccn*

Hall symbol: -P 2ab 2ac

$a = 18.292 (2) \text{ \AA}$

$b = 19.966 (2) \text{ \AA}$

$c = 7.8479 (4) \text{ \AA}$

$V = 2866.1 (4) \text{ \AA}^3$

$Z = 8$

$F_{000} = 1216$

$D_x = 1.341 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 11352 reflections

$\theta = 3.0\text{--}27.5^\circ$

$\mu = 0.09 \text{ mm}^{-1}$

$T = 183 (2) \text{ K}$

Cuboid, colorless

$0.15 \times 0.1 \times 0.1 \text{ mm}$

### Data collection

Nonius KappaCCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 183(2) \text{ K}$

$\varphi$  and  $\omega$  scans

Absorption correction: none

11352 measured reflections

3270 independent reflections

1391 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.082$

$\theta_{\text{max}} = 27.5^\circ$

$\theta_{\text{min}} = 3.0^\circ$

$h = -23 \rightarrow 23$

$k = -25 \rightarrow 25$

$l = -8 \rightarrow 10$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.082$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0346P)^2 + 0.003P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$S = 0.74$   $(\Delta/\sigma)_{\max} = 0.001$   
 3270 reflections  $\Delta\rho_{\max} = 0.14 \text{ e } \text{Å}^{-3}$   
 199 parameters  $\Delta\rho_{\min} = -0.20 \text{ e } \text{Å}^{-3}$   
 Primary atom site location: structure-invariant direct methods Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\text{sigma}(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.06301 (7)	0.63509 (6)	0.91292 (14)	0.0458 (3)
N1	0.07867 (7)	0.52183 (7)	0.88728 (15)	0.0363 (4)
C1	0.14376 (10)	0.52718 (10)	0.9744 (2)	0.0416 (5)
H1	0.1614	0.5668	1.0271	0.062*
C2	0.17911 (10)	0.46711 (9)	0.9736 (2)	0.0449 (5)
H2	0.2250	0.4575	1.0246	0.067*
C3	0.13464 (10)	0.42195 (9)	0.8828 (2)	0.0439 (5)
H3	0.1454	0.3761	0.8625	0.066*
C4	0.07299 (10)	0.45515 (9)	0.8283 (2)	0.0376 (4)
C5	0.01475 (10)	0.42109 (9)	0.73953 (18)	0.0385 (4)
H5	0.0220	0.3749	0.7156	0.058*
N2	-0.04575 (8)	0.44677 (7)	0.68969 (16)	0.0374 (4)
N3	-0.01941 (7)	0.57710 (7)	0.75188 (16)	0.0379 (4)
H3N	-0.0357	0.5364	0.7318	0.057*
C6	-0.09948 (9)	0.40501 (9)	0.61349 (19)	0.0356 (4)
C7	-0.14260 (10)	0.43399 (9)	0.4869 (2)	0.0407 (5)
H7	-0.1360	0.4797	0.4575	0.061*
C8	-0.19501 (10)	0.39616 (9)	0.4044 (2)	0.0416 (5)
H8	-0.2235	0.4157	0.3162	0.062*
C9	-0.20604 (10)	0.33021 (9)	0.4496 (2)	0.0447 (5)
H9	-0.2422	0.3044	0.3927	0.067*
C10	-0.16424 (10)	0.30146 (9)	0.5786 (2)	0.0466 (5)
H10	-0.1719	0.2560	0.6100	0.070*
C11	-0.11140 (10)	0.33934 (9)	0.6612 (2)	0.0415 (5)
H11	-0.0834	0.3201	0.7506	0.062*
C12	0.03966 (10)	0.58330 (9)	0.8518 (2)	0.0380 (4)
C13	-0.05737 (10)	0.63125 (8)	0.67646 (19)	0.0365 (4)

## supplementary materials

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C14	-0.13024 (10)	0.62156 (9)	0.6346 (2)	0.0440 (5)
H14	-0.1543	0.5814	0.6670	0.066*
C15	-0.16813 (11)	0.67072 (9)	0.5450 (2)	0.0501 (5)
H15	-0.2180	0.6640	0.5160	0.075*
C16	-0.13333 (12)	0.72922 (9)	0.4984 (2)	0.0507 (5)
H16	-0.1590	0.7626	0.4361	0.076*
C17	-0.06071 (11)	0.73919 (9)	0.5428 (2)	0.0465 (5)
H17	-0.0369	0.7797	0.5116	0.070*
C18	-0.02248 (11)	0.69054 (8)	0.63223 (19)	0.0410 (5)
H18	0.0271	0.6977	0.6630	0.061*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0529 (9)	0.0375 (8)	0.0470 (7)	-0.0021 (6)	-0.0039 (6)	-0.0077 (6)
N1	0.0378 (9)	0.0349 (9)	0.0362 (7)	-0.0004 (7)	0.0002 (7)	-0.0013 (7)
C1	0.0363 (12)	0.0492 (13)	0.0393 (10)	-0.0057 (10)	-0.0025 (8)	-0.0016 (8)
C2	0.0409 (12)	0.0450 (12)	0.0489 (11)	0.0015 (10)	-0.0008 (9)	0.0034 (9)
C3	0.0466 (12)	0.0370 (11)	0.0480 (11)	0.0026 (10)	0.0026 (9)	-0.0003 (9)
C4	0.0405 (12)	0.0360 (11)	0.0362 (9)	-0.0023 (10)	0.0037 (9)	0.0006 (8)
C5	0.0470 (12)	0.0343 (10)	0.0342 (9)	-0.0039 (10)	0.0070 (8)	-0.0009 (8)
N2	0.0385 (10)	0.0370 (9)	0.0369 (8)	-0.0013 (8)	0.0025 (7)	0.0001 (6)
N3	0.0394 (9)	0.0315 (8)	0.0428 (8)	-0.0016 (8)	-0.0021 (7)	-0.0018 (7)
C6	0.0389 (11)	0.0332 (11)	0.0346 (9)	-0.0028 (9)	0.0037 (8)	-0.0026 (8)
C7	0.0444 (12)	0.0361 (11)	0.0415 (10)	-0.0021 (9)	0.0068 (9)	0.0028 (8)
C8	0.0400 (12)	0.0472 (13)	0.0377 (9)	-0.0008 (9)	0.0022 (9)	0.0018 (9)
C9	0.0445 (13)	0.0473 (13)	0.0423 (10)	-0.0082 (10)	0.0017 (9)	-0.0043 (9)
C10	0.0550 (14)	0.0397 (11)	0.0450 (10)	-0.0087 (10)	0.0007 (9)	0.0019 (9)
C11	0.0461 (13)	0.0391 (12)	0.0394 (9)	-0.0039 (10)	0.0013 (9)	-0.0001 (8)
C12	0.0412 (12)	0.0369 (11)	0.0358 (10)	-0.0005 (10)	0.0053 (8)	-0.0004 (9)
C13	0.0447 (13)	0.0302 (11)	0.0345 (9)	0.0038 (9)	0.0046 (9)	-0.0020 (8)
C14	0.0431 (12)	0.0393 (12)	0.0497 (11)	-0.0012 (10)	0.0004 (9)	0.0020 (9)
C15	0.0503 (14)	0.0427 (13)	0.0573 (12)	0.0033 (10)	-0.0048 (10)	0.0028 (9)
C16	0.0669 (16)	0.0387 (12)	0.0465 (11)	0.0071 (10)	-0.0047 (10)	0.0030 (9)
C17	0.0648 (15)	0.0333 (11)	0.0415 (10)	-0.0014 (10)	0.0027 (9)	0.0005 (8)
C18	0.0496 (13)	0.0389 (12)	0.0344 (9)	-0.0027 (10)	0.0031 (8)	-0.0045 (8)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

O1—C12	1.2172 (19)	C7—H7	0.9500
N1—C1	1.377 (2)	C8—C9	1.379 (2)
N1—C4	1.413 (2)	C8—H8	0.9500
N1—C12	1.447 (2)	C9—C10	1.393 (2)
C1—C2	1.363 (2)	C9—H9	0.9500
C1—H1	0.9500	C10—C11	1.388 (2)
C2—C3	1.408 (2)	C10—H10	0.9500
C2—H2	0.9500	C11—H11	0.9500
C3—C4	1.376 (2)	C13—C14	1.386 (2)
C3—H3	0.9500	C13—C18	1.389 (2)

C4—C5	1.443 (2)	C14—C15	1.392 (2)
C5—N2	1.281 (2)	C14—H14	0.9500
C5—H5	0.9500	C15—C16	1.380 (2)
N2—C6	1.421 (2)	C15—H15	0.9500
N3—C12	1.341 (2)	C16—C17	1.388 (2)
N3—C13	1.415 (2)	C16—H16	0.9500
N3—H3N	0.8800	C17—C18	1.388 (2)
C6—C11	1.381 (2)	C17—H17	0.9500
C6—C7	1.394 (2)	C18—H18	0.9500
C7—C8	1.382 (2)		
C1—N1—C4	107.41 (14)	C8—C9—C10	120.03 (17)
C1—N1—C12	117.15 (14)	C8—C9—H9	120.0
C4—N1—C12	134.41 (14)	C10—C9—H9	120.0
C2—C1—N1	109.87 (16)	C11—C10—C9	119.82 (17)
C2—C1—H1	125.1	C11—C10—H10	120.1
N1—C1—H1	125.1	C9—C10—H10	120.1
C1—C2—C3	106.96 (17)	C6—C11—C10	120.05 (17)
C1—C2—H2	126.5	C6—C11—H11	120.0
C3—C2—H2	126.5	C10—C11—H11	120.0
C4—C3—C2	108.79 (17)	O1—C12—N3	126.29 (17)
C4—C3—H3	125.6	O1—C12—N1	118.13 (16)
C2—C3—H3	125.6	N3—C12—N1	115.58 (16)
C3—C4—N1	106.96 (15)	C14—C13—C18	120.09 (17)
C3—C4—C5	121.87 (17)	C14—C13—N3	117.68 (15)
N1—C4—C5	131.01 (16)	C18—C13—N3	122.03 (16)
N2—C5—C4	126.63 (17)	C13—C14—C15	120.00 (18)
N2—C5—H5	116.7	C13—C14—H14	120.0
C4—C5—H5	116.7	C15—C14—H14	120.0
C5—N2—C6	119.43 (15)	C16—C15—C14	120.08 (19)
C12—N3—C13	124.72 (15)	C16—C15—H15	120.0
C12—N3—H3N	117.6	C14—C15—H15	120.0
C13—N3—H3N	117.6	C15—C16—C17	119.77 (18)
C11—C6—C7	119.87 (16)	C15—C16—H16	120.1
C11—C6—N2	123.53 (15)	C17—C16—H16	120.1
C7—C6—N2	116.59 (15)	C18—C17—C16	120.60 (18)
C8—C7—C6	119.98 (16)	C18—C17—H17	119.7
C8—C7—H7	120.0	C16—C17—H17	119.7
C6—C7—H7	120.0	C17—C18—C13	119.44 (18)
C9—C8—C7	120.20 (17)	C17—C18—H18	120.3
C9—C8—H8	119.9	C13—C18—H18	120.3
C7—C8—H8	119.9		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N3—H3N $\cdots$ N2	0.88	1.83	2.691 (2)	166

Fig. 1

